# OBSERVATIONS & RECOMMENDATIONS

After reviewing data collected from **Dublin Lake**, **Dublin**, the program coordinators have made the following observations and recommendations.

Thank you for your continued hard work sampling the lake this year! Your monitoring group sampled the deep spot **three** times and the tributaries multiple times this year! As you know, conducting multiple sampling events each year enables DES to more accurately detect water quality changes. Keep up the great work!

## FIGURE INTERPRETATION

#### CHLOROPHYLL-A

Figure 1 and Table 1: Figure 1 in Appendix A depicts the historical and current year chlorophyll-a concentration in the water column. Table 1 in Appendix B lists the minimum, maximum, and mean concentration for each year that the lake has been monitored through VLAP.

Chlorophyll-a, a pigment found in plants, is an indicator of the algal abundance. Algae (also known as phytoplankton) are typically microscopic, chlorophyll producing plants that naturally occur in lake ecosystems. The chlorophyll-a concentration measured in the water gives biologists an estimation of the algal concentration or lake productivity. The median summer chlorophyll-a concentration for New Hampshire's lakes and ponds is 4.58 mg/m³.

The current year data (the top graph) show that the chlorophyll-a concentration *remained stable* from **June** to **July**, and then *increased* from **July** to **August**.

The historical data (the bottom graph) show that the **2010** chlorophyll-a mean is *much less than* the state median and is *slightly less than* the similar lake median. For more information on the similar lake median, refer to Appendix F.

Overall, visual inspection of the historical data trend line (the bottom graph) shows a **stable** in-lake chlorophyll-a trend since monitoring began. Specifically the mean chlorophyll concentration has **remained approximately the same** since **2007**.

After 10 consecutive years of sample collection, we will be able to conduct a statistical analysis of the historical data to objectively determine if there has been a significant change in the annual mean chlorophyll-a concentration since monitoring began.

While algae are naturally present in all lakes and ponds, an excessive or increasing amount of any type is not welcomed. In freshwater lakes and ponds, phosphorus is the nutrient that algae typically depend upon for growth in New Hampshire lakes. Algal concentrations may increase as nonpoint sources of phosphorus from the watershed increase, or as in-lake phosphorus sources increase. Therefore, it is extremely important for volunteer monitors to continually educate all watershed residents about management practices that can be implemented to minimize phosphorus loading to surface waters.

#### TRANSPARENCY

Figure 2 and Tables 3a and 3b: Figure 2 in Appendix A shows the historical and current year data for transparency with and without the use of a viewscope. Table 3a in Appendix B lists the minimum, maximum and mean transparency data without the use of a viewscope and Table 3b lists the minimum, maximum and mean transparency data with the use of a viewscope for each year that the lake has been monitored through VLAP.

Volunteer monitors use the Secchi disk, a 20 cm disk with alternating black and white quadrants, to measure how far a person can see into the water. Transparency, a measure of water clarity, can be affected by the amount of algae and sediment in the water, as well as the natural lake color of the water. **The median summer transparency for New Hampshire's lakes and ponds is 3.2 meters.** 

The current year data (the top graph) show that the non-viewscope inlake transparency *decreased* from **June** to **July**. A Secchi disk was not provided for the August sampling and we apologize for the inconvenience.

The historical data (the bottom graph) show that the **2010** mean non-viewscope transparency is *much greater than* the state median and

is **slightly less than** the similar lake median. Please refer to Appendix F for more information about the similar lake median.

The current year data (the top graph) show that the viewscope in-lake transparency *decreased* from **June** to **July**. The transparency measured with the viewscope was generally *greater than* the transparency measured without the viewscope this summer. A comparison of the transparency readings taken with and without the use of a viewscope shows that the viewscope typically increases the depth to which the Secchi disk can be seen into the lake, particularly on sunny and windy days. We recommend that your group measure Secchi disk transparency with and without the viewscope on each sampling event.

It is important to note that viewscope transparency data are not compared to a New Hampshire median or similar lake median. This is because lake transparency with the use of a viewscope has not been historically measured by DES. At some point in the future, the New Hampshire and similar lake medians for viewscope transparency will be calculated and added to the appropriate graphs.

Overall, visual inspection of the historical data trend line (the bottom graph) shows a **slightly increasing** trend for in-lake non-viewscope transparency, meaning that the transparency has **improved** since monitoring began in **2007**.

Again, please keep in mind that this trend is based on only *four* years of data. As previously discussed, after 10 consecutive years of sample collection, we will be able to conduct a statistical analysis of the historical data to objectively determine if there has been a significant change in the annual mean transparency since monitoring began.

Typically, high intensity rainfall causes sediment-laden stormwater runoff to flow into surface waters, thus increasing turbidity and decreasing clarity. Efforts to stabilize stream banks, lake and pond shorelines, disturbed soils within the watershed, and especially dirt roads located immediately adjacent to the edge of tributaries and the lake or pond should continue on an annual basis. Guides to best management practices that can be implemented to reduce, and possibly even eliminate, nonpoint source pollutants, are available from DES upon request.

## **TOTAL PHOSPHORUS**

Figure 3 and Table 8: The graphs in Figure 3 in Appendix A show the amount of epilimnetic (upper layer) phosphorus and hypolimnetic (lower layer) phosphorus; the inset graphs show current year data.

Table 8 in Appendix B lists the annual minimum, maximum, and median concentration for each deep spot layer and each tributary since the lake has been sampled through VLAP.

Phosphorus is typically the limiting nutrient for vascular aquatic plant and algae growth in New Hampshire's lakes and ponds. Excessive phosphorus in a lake or pond can lead to increased plant and algal growth over time. The median summer total phosphorus concentration in the epilimnion (upper layer) of New Hampshire's lakes and ponds is 12 ug/L. The median summer phosphorus concentration in the hypolimnion (lower layer) is 14 ug/L.

The current year data for the epilimnion (the top inset graph) show that the phosphorus concentration **remained stable** from **June** to **August**.

The historical data show that the **2010** mean epilimnetic phosphorus concentration is *much less than* the state median and is *slightly less than* the similar lake median. Refer to Appendix F for more information about the similar lake median.

The current year data for the hypolimnion (the bottom inset graph) show that the phosphorus concentration *decreased* from **June** to **July**, and then *remained stable* from **July** to **August**.

The historical data show that the **2010** mean hypolimnetic phosphorus concentration is *much less than* the state median and is *slightly less than* the similar lake median. Please refer to Appendix F for more information about the similar lake median.

Overall, visual inspection of the historical data trend line for the epilimnion shows a *relatively stable* phosphorus trend. Specifically, the mean annual epilimnetic phosphorus concentration has *remained approximately the same* since monitoring began in **2007**.

Overall, visual inspection of the historical data trend line for the hypolimnion shows a *decreasing* phosphorus trend since monitoring began. Specifically the mean annual concentration has *improved* since monitoring began in **2007**.

One of the most important approaches to reducing phosphorus loading to a waterbody is to continually educate watershed residents about the watershed sources of phosphorus and how excessive phosphorus loading can negatively impact the ecology and the recreational, economical, and ecological value of lakes and ponds.

# TABLE INTERPRETATION

# > Table 2: Phytoplankton

Table 2 in Appendix B lists the current and historical phytoplankton and/or cyanobacteria observed in the lake. Specifically, this table lists the three most dominant phytoplankton and/or cyanobacteria observed in the sample and their relative abundance in the sample.

The dominant phytoplankton and/or cyanobacteria observed in the **July** sample were **Asterionella** (**Diatom**), **Uroglenopsis** (**Golden-Brown**), and **Chrysosphaerella** (**Golden-Brown**).

Phytoplankton populations undergo a natural succession during the growing season. Please refer to the "Biological Monitoring Parameters" section of this report for a more detailed explanation regarding seasonal plankton succession. Diatoms and golden-brown algae populations are typical in New Hampshire's less productive lakes and ponds.

# > Table 4: pH

Table 4 in Appendix B presents the in-lake and tributary current year and historical pH data.

pH is measured on a logarithmic scale of 0 (acidic) to 14 (basic). pH is important to the survival and reproduction of fish and other aquatic life. A pH below 6.0 typically limits the growth and reproduction of fish. A pH between 6.0 and 7.0 is ideal for fish. The median pH value for the epilimnion (upper layer) in New Hampshire's lakes and ponds is **6.6**, which indicates that the state surface waters are slightly acidic. For a more detailed explanation regarding pH, please refer to the "Chemical Monitoring Parameters" section of this report.

The mean pH at the deep spot this year ranged from **6.20** in the hypolimnion to **6.74** in the epilimnion, which means that the water is **slightly acidic**.

It is important to point out that the hypolimnetic (lower layer) pH was *lower (more acidic)* than in the epilimnion (upper layer). This increase in acidity near the lake bottom is likely due to the decomposition of organic matter and the release of acidic by-products into the water column.

Due to the abundance of granite bedrock in the state and acid deposition received from snowmelt, rainfall, and atmospheric particulates, there is little that can be feasibly done to effectively increase lake pH.

## > Table 5: Acid Neutralizing Capacity

Table 5 in Appendix B presents the current year and historical epilimnetic ANC for each year the lake has been monitored through VLAP.

Buffering capacity (ANC) describes the ability of a solution to resist changes in pH by neutralizing the acidic input. The median ANC value for New Hampshire's lakes and ponds is **4.8 mg/L**, which indicates that many lakes and ponds in the state are at least "moderately vulnerable" to acidic inputs. For a more detailed explanation about ANC, please refer to the "Chemical Monitoring Parameters" section of this report.

The mean acid neutralizing capacity (ANC) of the epilimnion (upper layer) was **4.5 mg/L**, which is **approximately equal to** the state median. In addition, this indicates that the lake is **moderately vulnerable** to acidic inputs.

## > Table 6: Conductivity

Table 6 in Appendix B presents the current and historical conductivity values for tributaries and in-lake data. Conductivity is the numerical expression of the ability of water to carry an electric current, which is determined by the number of negatively charged ions from metals, salts, and minerals in the water column. The median conductivity value for New Hampshire's lakes and ponds is **40.0 uMhos/cm**. For a more detailed explanation, please refer to the "Chemical Monitoring Parameters" section of this report.

The mean annual epilimnetic conductivity at the deep spot this year was **76.1 uMhos/cm**, which is **slightly greater than** the state median.

The conductivity *increased slightly* in the lake and some tributaries this year. Due to the dry weather conditions in 2010, many tributaries experienced low or no flow conditions. This can concentrate ions and minerals and contribute to elevated conductivity levels.

In addition, the in-lake conductivity is **slightly greater than** the state median. Typically, elevated conductivity indicates the influence

of pollutant sources associated with human activities. These sources include failed or marginally functioning septic systems, agricultural runoff, stormwater runoff, and road runoff which can contain road salt during the spring snow-melt. New development in the watershed can alter runoff patterns and expose new soil and bedrock areas, which could also contribute to increasing conductivity. In addition, natural sources, such as iron and manganese deposits in bedrock, can influence conductivity.

It is possible that de-icing materials applied to nearby roadways during the winter months may be influencing the conductivity in the lake. In New Hampshire, the most commonly used de-icing material is salt (sodium chloride).

A limited amount of chloride sampling was conducted during **2010**. Please refer to the discussion of **Table 13** for more information.

Therefore, we recommend that the **epilimnion** (upper layer) and tributaries continue to be sampled for chloride next year. This additional sampling may help us identify what areas of the watershed are contributing to the increasing in-lake conductivity.

Please note that the DES Limnology Center in Concord is able to conduct chloride analyses, free of charge. As a reminder, it is best to conduct chloride sampling in the spring as the snow is melting and during rain events.

## > Table 8: Total Phosphorus

Table 8 in Appendix B presents the current year and historical total phosphorus data for in-lake and tributary stations. Phosphorus is the nutrient that limits the ability of algae and aquatic plants to grow and reproduce. Please refer to the "Chemical Monitoring Parameters" section of this report for a more detailed explanation.

The phosphorus concentrations in Aldridge, Boat Landing, Cemetary Cove, Culvert East, Dublin Lake Club, Eaves, Fernlea Road, Highfield Road, Julie Crocker Culvert, Korpi Culvert, Latchis Cove, Loon Point East, Oak Hill East, Spencer, and Womens Club were *relatively low* this year, which is good news. Due to the lack of rainfall in 2010, many waterbodies experienced lower phosphorus levels from decreased stormwater runoff.

However, we recommend that your monitoring group sample the major tributaries to the lake during snow-melt and periodically during rainstorms to determine if the phosphorus concentration is *elevated* in the tributaries during these times. Typically, the majority of nutrient loading to a lake occurs in the spring during snow-melt

and during intense rainstorms that cause soil erosion and surface runoff and within the watershed.

The total phosphorus concentration was *elevated* (95 ug/L) in **Latchis Beach East** this year. The total phosphorus concentration was also *elevated* (41 ug/L) in **Skyleau** on the **June** sampling event. We recommend that your monitoring group conduct a stream survey and rain event sampling along these tributaries so that we can determine what may be causing the elevated concentrations.

For a detailed explanation on how to conduct rain event sampling and stream surveys, please refer to the 2002 VLAP Annual Report special topic article, which is posted on the VLAP website at <a href="http://www.des.nh.gov/organization/divisions/water/wmb/vlap/categories/publications.htm">http://www.des.nh.gov/organization/divisions/water/wmb/vlap/categories/publications.htm</a>, or contact the VLAP Coordinator.

The total phosphorus concentration in **Merryman Rd.** was **slightly elevated** (**29 and 30 ug/L**) on the **June and October** sampling events. The turbidity of the samples was also **elevated** (**13.2 and 15.2 NTUs**). The total phosphorus concentration in **Oak Hill** was **slightly elevated** (**26 ug/L**) on the June sampling event, and the turbidity was also **slightly elevated** (**3.09 NTUs**).

This suggests that the stream bottom may have been disturbed while sampling or that erosion is occurring in the watershed. When the stream bottom is disturbed, phosphorus rich sediment is released into the water column. When collecting tributary samples, please be sure to sample where the tributary is flowing and where the stream is deep enough to collect a "clean" sample free from organic debris and sediment.

If you suspect that erosion is occurring in this area of the watershed, we recommend that your monitoring group conduct a stream survey and rain event sampling along this tributary. This additional sampling may allow us to determine what is causing the *elevated* levels of turbidity and phosphorus.

For a detailed explanation on how to conduct rain event sampling and stream surveys, please refer to the 2002 VLAP Annual Report special topic article, which is posted on the VLAP website at <a href="http://www.des.nh.gov/organization/divisions/water/wmb/vlap/categories/publications.htm">http://www.des.nh.gov/organization/divisions/water/wmb/vlap/categories/publications.htm</a>, or contact the VLAP Coordinator.

Table 9 and Table 10: Dissolved Oxygen and Temperature Data Table 9 in Appendix B shows the dissolved oxygen/temperature profile(s) collected during **2010**. Table 10 in Appendix B shows the

historical and current year dissolved oxygen concentration in the hypolimnion (lower layer). The presence of sufficient amounts of dissolved oxygen in the water column is vital to fish and amphibians and bottom-dwelling organisms. Please refer to the "Chemical Monitoring Parameters" section of this report for a more detailed explanation.

The dissolved oxygen concentration was **high** at all deep spot depths sampled at the lake on the **July** sampling event. As thermally stratified lakes age, and as the summer progresses, oxygen typically becomes **depleted** in the hypolimnion (lower layer) by bacterial decomposition. Specifically, the loss of oxygen in the hypolimnion results primarily from biological organisms using oxygen to break down organic matter, both in the water column and particularly at the bottom of the lake or pond where the water meets the sediment. The **high** oxygen level in the hypolimnion is a sign of the lake's overall good health. We hope this continues!

The dissolved oxygen concentration was greater than **100 percent** saturation between **7.0** and **11.0** meters at the deep spot on the **July** sampling event. Wave action from wind can dissolve atmospheric oxygen into the upper layers of the water column. Layers of algae can also increase the dissolved oxygen in the water column, since oxygen is a by-product of photosynthesis. Considering that the depth of sunlight penetration into the water column was approximately **8.0** meters on this sampling event, as shown by the viewscope Secchi disk transparency depth, and that the metalimnion, the layer of rapid decrease in water temperature and increase in water density where algae typically congregate, was located between approximately **7.0** and **11.0** meters, we suspect that an abundance of algae in the metalimnion caused the oxygen super-saturation.

## > Table 11: Turbidity

Table 11 in Appendix B lists the current year and historical data for in-lake and tributary turbidity. Turbidity in the water is caused by suspended matter, such as clay, silt, and algae. Water clarity is strongly influenced by turbidity. Please refer to the "Other Monitoring Parameters" section of this report for a more detailed explanation.

The turbidities in the Aldridge, Culvert East, Eaves, Fernlea Road, Merryman Rd., and Oak Hill samples were *elevated* (2.54, 3.91, 2.15, 7.89, 13.2, and 3.09 NTUs) on the June sampling event.

The turbidities in the Culvert East and Merryman Rd. samples were *elevated* (2.35 and 15.2 NTUs) on the October sampling event.

The dry weather conditions in 2010 suggest that the streams experienced low or no flow conditions. This suggests the bottom may have been disturbed while sampling. When the stream bottom is disturbed, sediment, which typically contains attached phosphorus, is released into the water column. When collecting tributary samples please sample where there's sufficient stream flow and depth to collect a "clean" sample free from debris and sediment.

# > Table 12: Bacteria (E.coli)

Table 12 in Appendix B lists the current year and historical data for bacteria (E.coli) testing. E. coli is a normal bacterium found in the large intestine of humans and other warm-blooded animals. E.coli is used as an indicator organism because it is easily cultured and its presence in the water, in defined amounts, indicates that sewage **may** be present. If sewage is present in the water, potentially harmful disease-causing organisms **may** also be present.

The *E.coli* concentration was **low** on each sampling event at the **Boat** Landing, Cemetary Cove, Culvert East, Dublin Lake Club, Eaves, Fernlea Road, Highfield Road, Julie Crocker Culvert, Korpi Culvert, Latchis Beach, Latchis Cove, Loon Point East, Merryman Rd., Oak Hill, Oak Hill East, Outlet Spencer, Skyleau, and Womens Club this year. We hope this trend continues!

If residents are concerned about sources of bacteria, such as failing septic systems, animal waste, or waterfowl waste, it is best to conduct *E. coli* testing when the water table is high, when beach use is heavy, or immediately after rain events.

The **Aldridge** *E. coli* concentration was *elevated* on the **6/3/2010** sampling event. The concentration of **670** counts per 100 mL *was greater than* the state standard of 406 counts per 100 mL for recreational waters that are not designated public beaches. *E. coli* concentrations decreased to **20** counts per 100 mL on the **6/14/2010** sampling event and were also low on the **10/12/2010** sampling event.

If concentrations continue to be elevated in future years, we recommend that your monitoring group conduct rain event sampling and bracket sampling in this area to help determine the bacteria source.

For a detailed explanation on how to conduct rain event sampling and stream surveys, please refer to the 2002 VLAP Annual Report special topic article, which is posted on the VLAP website at

http://www.des.nh.gov/organization/divisions/water/wmb/vlap/categories/publications.htm, or contact the VLAP Coordinator.

## > Table 13: Chloride

Table 13 in Appendix B lists the current year and the historical data for chloride sampling. The chloride ion (Cl-) is found naturally in some surfacewaters and groundwaters and in high concentrations in seawater. Research has shown that elevated chloride levels can be toxic to freshwater aquatic life. In order to protect freshwater aquatic life in New Hampshire, the state has adopted **acute and chronic** chloride criteria of **860 and 230 mg/L** respectively. The chloride content in New Hampshire lakes is naturally low, generally less than 2 mg/L in surface waters located in remote areas away from habitation. Higher values are generally associated with salted highways and, to a lesser extent, with septic inputs. Please refer to the "Chemical Monitoring Parameters" section of this report for a more detailed explanation.

The **epilimnion** was sampled for chloride during the **June**, **July and August** sampling events. The results were **13 and 14 mg/L**, which are *much less than* the state acute and chronic chloride criteria. However, this concentration is *greater than* what we would normally expect to measure in undisturbed New Hampshire surface waters.

We recommend that your monitoring group continue to conduct chloride sampling in the epilimnion, particularly in the spring during snow-melt and rain events during the summer. This will establish a baseline of data that will assist your monitoring group and DES to determine lake quality trends in the future.

Please note that chloride analyses can be run free of charge at the DES Limnology Center. Please contact the VLAP Coordinator if you are interested in chloride monitoring. In addition, it is best to conduct chloride sampling in the spring as the snow is melting and during rain events.

➤ Table 14: Current Year Biological and Chemical Raw Data
Table 14 in Appendix B lists the most current sampling year results.

Since the maximum, minimum, and annual mean values for each parameter are not shown on this table, this table displays the current year "raw," meaning unprocessed, data. The results are sorted by station, depth, and then parameter.

## Table 15: Station Table

As of the spring of 2004, all historical and current year VLAP data are included in the DES Environmental Monitoring Database (EMD). To facilitate the transfer of VLAP data into the EMD, a new station identification system had to be developed. While volunteer monitoring groups can still use the sampling station names that they have used in the past and are most familiar with, an EMD station name also exists for each VLAP sampling location. Table 15 in Appendix B identifies what EMD station name corresponds to the station names you have used in the past and will continue to use in the future.

# **DATA QUALITY ASSURANCE AND CONTROL**

#### **Annual Assessment Audit:**

During the annual visit to your lake, the biologist conducted a sampling procedures assessment audit for your monitoring group. Specifically, the biologist observed the performance of your monitoring group and completed an assessment audit sheet to document the volunteer monitors' ability to follow the proper field sampling procedures, as outlined in the VLAP Monitor's Field Manual. This assessment is used to identify any aspects of sample collection in which volunteer monitors failed to follow proper procedures, and also provides an opportunity for the biologist to retrain the volunteer monitors as necessary. This will ultimately ensure samples that the volunteer monitors collect are truly representative of actual lake and tributary conditions.

Overall, your monitoring group did an *excellent* job collecting samples on the annual biologist visit this year! Specifically, the members of your monitoring group followed the proper field sampling procedures and there was no need for the biologist to provide additional training. Keep up the good work!

# Sample Receipt Checklist:

Each time your monitoring group dropped off samples at the laboratory this summer, the laboratory staff completed a sample receipt checklist to assess and document if your group followed proper sampling techniques when collecting the samples. The purpose of the sample receipt checklist is to minimize, and hopefully eliminate, improper sampling techniques.

Overall, the sample receipt checklist showed that your monitoring group did an *excellent* job when collecting samples and submitting them to the laboratory this year! Specifically, the members of your monitoring group

followed the proper field sampling procedures and there was no need for the laboratory staff to contact your group with questions, and no samples were rejected for analysis.

## **USEFUL RESOURCES**

Acid Deposition Impacting New Hampshire's Ecosystems, DES fact sheet ARD-32, (603) 271-2975 or

www.des.nh.gov/organization/commissioner/pip/factsheets/ard/documents/ard-32.pdf.

Best Management Practices to Control Nonpoint Source Pollution: A Guide for Citizens and Town Officials, DES Booklet WD-03-42, (603) 271-2975 or

www.des.nh.gov/organization/commissioner/pip/publications/wd/docu ments/wd-03-42.pdf.

Erosion Control for Construction in the Protected Shoreland Buffer Zone, DES fact sheet WD-SP-1, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/factsheets/sp/documents/sp-1.pdf

Lake or Pond – What is the Difference? DES fact sheet WD-BB-49, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/factsheets/bb/documents/bb-49.pdf

Low Impact Development Hydrologic Analysis. Manual prepared by Prince George's County, Maryland, Department of Environmental Resources. July 1999. To access this document, visit www.epa.gov/owow/nps/lid\_hydr.pdf or call the EPA Water Resource Center at (202) 566-1736.

Low Impact Development: Taking Steps to Protect New Hampshire's Surface Waters, DES fact sheet WD-WMB-17, (603) 271-2975 or www.des.nh.gov/organization/commissioner/pip/factsheets/wmb/documents/wmb-17.pdf.

NH Stormwater Management Manual Volume 1: Stormwater and Antidegradation, DES fact sheet WD-08-20A, (603) 271-2975 or http://des.nh.gov/organization/commissioner/pip/publications/wd/doc uments/wd-08-20a.pdf

NH Stormwater Management Manual Volume 2: Post-Construction Best Management Practices Selection and Design, DES fact sheet WD-08-20B, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/publications/wd/doc uments/wd-08-20b.pdf

NH Stormwater Management Manual Volume 3: Erosion and Sediment Controls During Construction, DES fact sheet WD-08-20C, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/publications/wd/doc uments/wd-08-20c.pdf

Road Salt and Water Quality, DES fact sheet WD-WMB-4, (603) 271-2975 or

www.des.nh.gov/organization/commissioner/pip/factsheets/wmb/documents/wmb-4.pdf.

Sand Dumping - Beach Construction, DES fact sheet WD-BB-15, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/factsheets/bb/documents/bb-15.pdf.

Vegetation Maintenance Within the Protected Shoreland, DES fact sheet WD-SP-5, (603) 271-2975 or

http://des.nh.gov/organization/commissioner/pip/factsheets/sp/documents/sp-5.pdf